LINEAR SESTERTERPENES FROM THE CARIBBEAN SPONGE THORECTA HORRIDUS WITH INFLAMMATORY ACTIVITY

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Abstract: Two new linear sesterterpenes 1 and 2, containing an α,β -unsaturated γ -lactone ring, have been isolated from the Caribbean sponge *Thorecta horridus*, Hyatt 1877 (F. Thorectidae). The structures of the two new metabolites were established on the basis of spectral data including 2D NMR experiments. Compound 1 exhibits inflammatory activity.

Among the various classes of polyisoprenoids, sesterterpenes seem to be the most unusual metabolites from the marine environment². In fact, the majority of these relatively rare secondary metabolites have been isolated from marine invertebrates. As far as the structures of the marine-derived sesterterpenes is concerned, most of them are based on a polycarbocyclic skeleton, while a limited number of linear furanoid compounds are also known. All the isolated compounds appear to be biosynthesized through rather elaborated pathways starting from the common precursor geranylfarnesol. This elaboration also comprises drastic modifications with the loss or addition of one or more carbon units.

During our investigation into bioactive secondary metabolites from marine invertebrates we have identified in the Caribbean sponge *Thorecta horridus* two new linear sesterterpenes (1 and 2) which hypothetically derive from the precursor through a simple oxidation at C-1 and C-21 carbon atoms. These new compounds represent the closest analogues of geranylfarnesol which have been so far observed in marine sources. The most abundant compound 1 possesses inflammatory activity.

Specimens of sponge *Thorecta horridus*, Hyatt 1877 (F. Thorectidae) were collected by scuba at a depth of 30 m off Grand Bahama Island, during an expedition in July 1990. The MeOH/ Toluene (3:1) extract was initially fractionated by MPLC over silica-gel (230-400 mesh), with the final purification by RP-18 HPLC (100% MeOH) yielding compounds 1 and 2 (0.43 and 0.02 % dry weight of the sponge, respectively).

The molecular formula for the major compound 1, $C_{25}H_{38}O_2$, was determined from interpretation of the M⁺ (m/z 370.2878) ion in the HREIMS³ in conjunction with ¹H and ¹³C NMR data (Table 1). The molecular formula indicated 7 degrees of unsaturation. Ten olefinic resonances and 1 carbonyl resonance in the ¹³C NMR spectrum accounted for 6 unsaturations, leaving 1 ring in compound 1. The nature of this ring was suggested by the FT-IR bands at 1783 and 1749 cm⁻¹, characteristic of β -substituted α , β -unsaturated γ -lactones and by the ¹H NMR spectrum of 1 containing two long-range coupled (J = 1.3 Hz) signals at δ 4.73 (2H, d) and δ 5.85 (1H, quintet) attributable to the allylic methylene protons at C-21 and the lactone olefinic proton at C-2, respectively.

One- (spin-decoupling and nOe difference) and two-dimensional (COSY and long range COSY) ¹H NMR experiments allowed us to extend the lactone ring up to C-8. Useful information arose from long range couplings observed between H-2 and H₂-4, H-6 and H₃-22, H-6 and H₂-8, indicative of their allylic relationship and between the homoallylic H₂-5 and H₂-8, and from a nOe enhancement of the H₂-4 signal observed when H₂-21 was irradiated, which proved the spatial proximity of these two methylene groups.

The proposed substructure was further supported by the assignment of the resonance of the carbon atoms of the above fragment in the ¹³C NMR spectrum of 1 (Table 1) based on DEPT and 2D heteronuclear correlation via ¹J (HETCORR) and ^{2,3}J (COLOC). Particularly, the COLOC experiment displayed significant correlations between C-1 and

Table 1. 13C and ¹H NMR assignment for compounds 1 and 28

			plet																							
	H	4.76, 2H, bd (2.0)	7.10, 1H, narrow mult		2.32,2H, bt (7.5)	2.27,2H, bq (7.5)	5.121		2.07m	1.98 ⁿ	5.12		2.07™	1.98n	5.12		2.07m	1.98 ⁿ	5.12		1.68, 3H, s		1.59P	1.59P	1.59P	1.59P
•	3 61	70.10	144.0	137.25	25.45 ^b	25.69b	122.59	135.15^{c}	39.70 ^d	26.73e	124.35	134.86°	39.70 ^d	26.62°	124.14 ^f	134.86	39.70d	26.62	124.00	130.50	25.69	174.50	16.10	16.00	16.00	17.70
	DEPT	CH ₂	5	ပ	.	CH ₂	В	ບ	CH,	CH,	8	ບ	Œ,	CH,	Ð	ပ	CH_2	CH ₂	5	ပ	සී	၁	G,	GH3	£,	GH,
	н,		5.85, 1H, quintet(1.3)		2.45, 2H, bt (7.5)	2.30, 2H, bq (7.5)	5.128		2.07h	1.98	5.128		2.07 th	1.98	5.128		2.07h	1.98	5.128		1.68, 3H, s	4.73, 2H, d (1.3)	1.62, 3H, bs	1.59°	1.59	1.59°
- I	J81	170.17	115.55	135.28e	28.71	25.65	121.75	137.58	39.69p	26.71€	124.32 ^d	134.92°	39.64b	26.58°	124.11 ^d	134.92	39.57 ^b	26.43°	123.78 ^d	131.26	25.65	73.19	16.15	15.98	15.98	17.65
	DEPT	υ	ŧ	ບ	CH ²	CH,	B	ပ	CH ²	CH,	Э	ပ	CH ²	CH,	B	ပ	CH,	GH ²	£	υ	සී	CH_2	GH,	£	ቻ	.
	carbon	-	7	3	4	S	9	7	∞	6	10	11	12	13	41	15	91	17	18	19	8	21	23	23	24	22

Assignments determined by analogy to model compounds and by the 1D and 2D NMR experiments reported in the text. ¹H (500MHz) and ¹³C (125MHz) NMR spectra were recorded in CDCl₃ with TMS as internal standard. Proton coupling constants () are given in Hertz.

b-f Values with identical superscript within each column may be interchanged. 8-n Values with identical superscript within each column are mutually overlapped.

Broad singlet, 9H. P Broad singlet, 12H.

H-2, C-1 and H_2 -21, C-2 and H_2 -21, C-2 and H_2 -4, C-5 and H_2 -4, C-6 and H_3 -22. C-7 and H_3 -22.

The E-configuration of the C-6 double bond was inferred from the 13 C chemical shifts of the allylic methyl and methylenes and confirmed by the nOe enhancement of H_3 -22 on irradiation at δ 2.30 (H_2 -5).

The rest of the ¹H NMR spectrum of 1 consists of four methyl, five methylenes and three olefinic methine signals whose chemical shifts and multiplicities strongly suggest that an all-trans C-15 prenyl chain is linked to C-8. ¹³C NMR spectrum, which displays all the resonances for the pertinent carbon atoms, corroborates the presence of the polyprenyl chain with E-stereochemistry of the C-10 and C-14 double bonds (Table 1).

The minor component 2 possesses a molecular formula $C_{25}H_{38}O_2$ determined on the basis of HREIMS³ (m/z 370.2869), thus being an isomer of 1. Its structure was determined by direct comparison of the ¹H and ¹³C NMR spectra (Table 1) of the two compounds which illustrated that 2 also contained an all-trans polyprenyl chain with a terminal α,β -unsaturated γ -lactone ring.

However, there were relevant differences in the NMR resonances of the methine proton of the α,β -unsaturated lactone which are shifted downfield from δ 5.85 in 1 to δ 7.10 in 2 in the proton spectrum and from δ 115.55 in 1 to δ 144.0 in 2 in the carbon-13 spectrum. These data required that the methine proton must be β -positioned to the carbonyl function in 2. This was substantiated by the carbonyl band at 1754 cm⁻¹ in the FT-IR spectrum which did not display the characteristic splitting due to the presence of an α -proton.

The above data led us to the conclusion that the difference between the two isomers consisted in the position of the carbonyl and the oxymethylene groups which in 1 were located at C-1 and C-21, respectively whereas in 2 were inversely positioned.

An excellent confirmation of this assumption arose from the homoallylic relationship between H₂-1 and H₂-4, illustrated by an ¹H-¹H COSY long range experiment and from the observed chemical shifts of both ¹H and ¹³C signals of the methylene group at C-4, which were upfield-shifted when compared with the corresponding CH₂ in 1. An assignment of the resonances in the ¹³C NMR spectrum of 2, which were based on selective decouplings, DEPT experiments and comparison with model compounds supported the proposed structure.

Compound 1 exhibited inflammatory activity both in vivo (paw oedema) and in vitro (release of histamine) experiments. Injection of 0.5 - 1- 2 mg into the rat paw induced a dose-dependent oedema with a peak response occurring 1 h after the injection (Fig. A). When compound 1 was injected in rats pre-treated with mepyramine (2.5 mg/Kg) and methysergide (3 mg/Kg) the oedema induced by 1 mg compound 1 was suppressed by about 65% (Fig. A), suggesting that the inflammatory response was brought about by the release of histamine and/or 5-hydroxytryptamine. This was confirmed by the ability of 1 (0.1-1-10 µg/ml) to induce the release of histamine from rat peritoneal cells (Fig.B) in a

concentration-dependent fashion (4%, 11% and 30%). The degree of histamine release was not significantly affected by phosphatidylserine (Fig.B).

These results suggest that 1 is able to induce the release of histamine both *in vivo* and *in vitro* by a mechanism which is unaffected by phosphatidylserine and therefore possibly depends on mobilization of intracellular calcium stores.

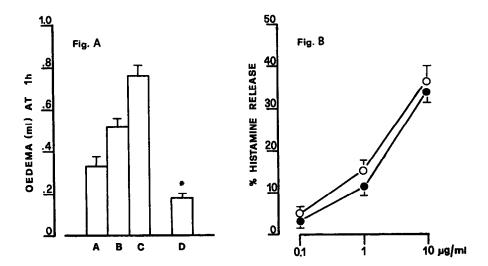


Fig. A. Oedema of the hind paw of male Wistar rats (120-140g) induced by subplantar injection of 0.5(A)-1(B)-2(C) mg compound 1. Inhibition (D) by mepyramine (2.5 mg/Kg) and methysergide (3 mg/Kg) both given intraperitoneally 30 min before the paw injection of 1 (1 mg). The volume of the paw was measured by plethismometer⁴ every 30 min for 2 h and compared to the initial one. Each bar represents the mean \pm s.e.m. for n = 5 rats of the oedema ccurring at 1 h. * Shows a significant difference from untrated rats (P < 0.05).

Fig.B. Effect of various concentrations of 1 on the histamine release from rat peritoneal cells in absence (solid circles) or presence (open circles) of phosphatidylserine (50 μ g/ml). Each point is an average of two triplicate experiments.⁵

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- 3. Additional spectral data : For 1, HREIMS m/z (assignment relative intensity) 370.2878 (M⁺, $C_{25}H_{38}O_2$ requires 370.2872, 26), 355.2621 (M⁺ Me, $C_{24}H_{35}$ O_2 , 1), 327.2322 (M⁺ 43 , $C_{22}H_{31}O_2$,7), 301.2222 (M⁺ 69 , $C_{20}H_{29}O_2$,1), 205.1973 (M⁺ -165 , $C_{15}H_{25}$, 10), 204.1890 (M⁺ 166 , $C_{15}H_{24}$, 2), 191.1075 (M + -179 , C_{12} H_{15} O_2 , 4), 166.0995 (M⁺ -204 , $C_{10}H_{14}$ O_2 , 22), 149.1336 (M⁺ -221 , $C_{11}H_{17}$, 13), 136.1256 (M⁺ -234 , $C_{10}H_{16}$, 25), 121.1015 (M⁺ -249 , $C_{9}H_{13}$, 16), 98.0336 (M⁺ 272 , $C_{5}H_{6}O_2$, 68), 69.0702 (M⁺ -301 , $C_{5}H_{9}$, 100) ; FT-IR (CHCl₃) 1783, 1749, 1639, 1447, 1435 cm -1 : For 2, HREIMS m/z (assignment relative intensity) 370.2869 (M⁺, $C_{25}H_{38}O_2$ requires 370.2872 , 29), 355.2621 (M⁺ Me , $C_{24}H_{35}$ O_2 , 2), 327.2322 (M⁺ 43 , $C_{22}H_{31}O_2$, 6), 301.2222 (M⁺- 69, $C_{20}H_{29}O_2$, 14), 205.1973 (M⁺ -165 , $C_{15}H_{25}$, 10), 204.1890 (M⁺ 166 , $C_{15}H_{24}$, 13), 191.1075 (M + -179 , C_{12} H_{15} O_2 , 16), 166.0995 (M⁺ -204 , $C_{10}H_{14}$ O_2 , 3), 165.0914 (M⁺ -205, $C_{10}H_{14}$ O_2 , 6), 149.1336 (M⁺ -221, $C_{11}H_{17}$, 13), 136.1256 (M⁺ -234 , $C_{10}H_{16}$, 32), 121.1015 (M⁺ 249, $C_{9}H_{13}$, 15), 98.0336 (M⁺ 272 , $C_{5}H_{6}O_2$, 11), 69.0702 (M⁺ -301 , $C_{5}H_{9}$, 100) ; FT-IR (CHCl₃) 1754, 1635, 1440, 1435 cm⁻¹.
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